

# THE USE OF DIFFERENTIAL SCANNING CALORIMETRY IN LIFE SCIENCES

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- Technical principles
- Design and optimisation of an experiment
- Different types of applications
- Troubleshooting



- A protein's function depends on its 3D-structure
- Loss of structural integrity with accompanying loss of activity is called denaturation
- Proteins could be denatured by:
  - heat or cold
  - pH extremes
  - organic solvents
  - chaotropic agents: urea and guanidinium hydrochloride



## METHODS TO MEASURE PROTEIN UNFOLDING

- Temperature induced denaturation by fluorescence, Circular Dichroism, Differential Scanning Calorimetry.
- Chemically-induced denaturation curves by fluorescence or Circular Dichroism



Protein stability could be modulated by external factors.

Co-solute could increase the stability (Tm) of a protein, then probable increase in shelf-life-formulations.

If a compound at stoichiometric concentrations increases Tm then we have specific binding.





- Differential Scanning Calorimetry (DSC) measures the temperatures and heat flows associated with transitions in materials as a function of time and temperature in a controlled atmosphere.
- These measurments provide qualitative and quantitative information about physical and chemical changes that involve exothermic or endothermic processes or changes in heat capacity.



- The DSC contains two sample cells:
  - One cell contains biomolecule (e.g. protein) in buffer
  - The other cell contains only the buffer
  - DSC cells are either capillary or "lollipop" in shape, and there are always two of them:





## **DSC : TECHNICAL PRINCIPLES**

The DSC cells are contained in an insulated "adiabatic" chamber.

The device is designed to maintain the two cells at the same temperature, as they are heated.

The DP is the differential power added to maintain  $\Delta T \sim 0$  between the cells (data collected in an experiment).





 Although the two cells in the DSC are manufactured to be as identical as possible, there will be slight differences in volume, shape, etc...





- Thermal transition midpoint Tm (melting temperature): Indication of thermal stability.
- Enthalpy △H: Includes energy associated with changes in inter- and intramolecular interactions (hydrogen bonds, etc.).
- **Entropy**  $\triangle$  S: "Molecular disorder"
- Heat capacity  $\triangle$  Cp: Measures ability of biomolecule to absorb heat energy without increase in temperature.
- **Gibbs free energy**  $\triangle G: \triangle G = \triangle H T \triangle S$ . At  $T_m, \triangle G_{unfolding} = 0$ .



### **DSC PRODUCTS**



VP-DSC: Cell volume 500 µL 4 experiments/8 hours

Capillary cells Cell Volume 160 µL Up to 50 experiments/day Unattempted operation Up to 576 samples on board



### **Concentration requirements:**

Depends on the molecular weight of the protein

Minimum concentration 0.02 mg/ml

As starting point min 0.1-0.2 mg/ml

Maximum concentration 50 - 100 mg/ml



### **Sample preparation:**

Exchange material into buffer using dialysis or desalting column

Retain the exchange buffer for use as the reference solution

Centrifuge or filter sample



### **Choosing a good buffer:**

Compatible with many buffers

Avoid DTT (Unstable and undergoes oxidation)

Use  $\beta$  -mercaptoethanol or TCEP

Tris buffer should not be used

First, try DSC buffer only



- Both cells are loaded with buffer
- The instrument is setup for multiple (20) data collection runs (heating/cooling cycles)
- "Buffer/buffer" data is collected ( $\geq$ 3 runs)
- When the instrument is cooling down, prior to a heating cycle, the protein is introduced at 25 °C
- A "protein/buffer" data run is collected



Temp (T)

Temp (T)

Temp (T)



## **PRINCIPLES OF ANALYSIS**



















### Optimization of purification conditions



Higher yield in protein purification



Solubility and stability study of antigens for the elaboration of a potential vaccine candidate





### Stability screening for drug formulations development







### Protein could loose activity upon prolonged storage





#### Rank order binding study - High throughput technique

Identification of a Chemoreceptor for Tricarboxylic Acid Cycle Intermediates





### **Protein interactions**



Using specific fitting procedure, a value of Kd could be approximately determined



Differential scanning calorimetry (DSC) of the DNA/lipid complex at different molar ratios.



Pector et al., J. Biol. Chem., 2000



#### Adenylation-induced structural changes in NAD+-DNA ligases





### **Differential Scanning Proteolysis**



Pey Biochim. et Biophys. acta, 2013



### Thermodynamic parameters of aggregates dissolution

Amyloid fibrils are involved in neurodegenerative diseases





Morel et al., J. Phys Chem B, 2010



## TROUBLESHOOTING

### Why do DSC experiments not work as expected?

- Incorrect sample preparation
- Protein already denatured prior to DSC
- Incorrect concentrations used
- Buffer mismatch between reference and sample cells
- Incorrect filling technique
- "Thermal history" not established



## **PROBLEMS IN DSC DATA COLLECTION**

• Air bubbles displace liquid and therefore reduce the heat capacity (yielding erroneous results).



To address this issue:

- Samples & buffer are degassed (10 min)
- DSC cell is kept under pressure (~35 psi)
- A certain technique is used in filling the cells



### Pressure changes affect the apparent heat capacity



This is not a critical issue to derive thermodynamic parameters



## CONCLUSIONS

- Ideal for stability and folding studies
- Identify conditions that guarantee long term stability
- Ideal to identify ligands of unknown proteins
- Monitor reversibility of thermal processes
- Study molecules in their native state without labeling
- Can be use with solutions that interfere with optical methods including turbid or colored solutions or particulate suspensions
- Monitor conformational energetics of proteins and biopolymers